Molecular dynamics simulation

A numerical technique to compute the equilibrium and transport properties of classical many-particle systems by solving Newton's equation of motion.

$$H = \sum_{i=1}^{N} \frac{p_i^2}{2m} + V_{\text{pot}}(r_1, \dots, r_N)$$



van der Waals system

van der Waals bonding - fluctuating electric dipole moments

$$V_{\text{pot}}(\boldsymbol{r}_1,\ldots,\boldsymbol{r}_N) = \sum_i \sum_{j>i} v_2(\mid \boldsymbol{r}_i - \boldsymbol{r}_j \mid)$$

Model for rare gas solids

Ex. Lennard-Jones potential (the generic model potential in MD)

$$v_2(r_{ij}) = 4\epsilon \left[\left(\frac{\sigma}{r_{ij}} \right)^{12} - \left(\frac{\sigma}{r_{ij}} \right)^6 \right]$$
$$r_{ij} \equiv \mid \mathbf{r}_i - \mathbf{r}_j \mid$$



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Metallic systems

Metallic bonding - delocalization of valence electrons

Physical fact: Pairwise interaction fails. The bonds become weaker when the local environment becomes more crowded.

Pair-functional model

$$V_{\text{pot}}(\boldsymbol{r}_1, \dots, \boldsymbol{r}_N) = \sum_i g_i(\rho_i) + \sum_i \sum_{j>i} v_2(|\boldsymbol{r}_i - \boldsymbol{r}_j|)$$

with

$$ho_i = \sum_{j
eq i} h_i (\mid r_i - r_j \mid)$$



6N-dim. phase-space trajectory



Instability of trajectories

Lyaponov instability

 $oldsymbol{r}_i(t)$ - parent trajectory

 $r_i^\prime(t)$ - perturbed trajectory

The initial velocity for two particle is changed slightly to obtain the perturbed trajectory

$$v'_{x1}(0) = v_{x1}(0)[1+10^{-10}]$$

 $v'_{x2}(0) = v'_{x2}(0)[1-10^{-10}]$

The following deviation is determined

$$\Delta(t) = \sum_{i=1}^{N} [r_i(t) - r'_i(t)]^2$$

Liquid Argon, N=1000



Time evolution



A typical program



Equilibration



Periodic boundary conditions



- the simulation cell is replicated throughout space to form an infinite lattice
- only particles in the central box are stored

Periodic boundary conditions



Assume short range interaction: $r_c < L/2$

This implies interaction with only the nearest image

Periodic boundary conditions



Minimum image convention

$$egin{aligned} & r_{ij} = r_j - r_i \ & r_{ij} = r_{ij} - L \left[r_{ij} / L
ight] \end{aligned}$$

where

[X] is the nearest integer to X

Example of average quantities

Simple averages
$$\langle A
angle$$

- temperature
- pressure

Fluctuations $\left\langle \left(\delta A \right)^2 \right\rangle$

heat capacity

Static quantities

- $\langle AB \rangle$
- pair-distribution function
- static structure factor

Dynamic quantities $\langle A(t)B(0)\rangle$

- mean squared displacement
- velocity correlation function
- diffusion coefficient
- dynamic structure factor

Pair distribution function



Static structure factor



Velocity correlation function



Velocity correlation function – spectral function



Mean squared displacement

