General soft matter

- 1. Discuss the four unifying concepts of soft matter.
- 2. What is a viscoelastic material? Explain what is meant by shear thinning and shear thickening.

Phase transitions

- 3. What is the free energy of mixing within the regular solution model?
- 4. Define the coexistence curve or binodal.
- 5. Define the spinodal.
- 6. What is meant by uphill diffusion?
- 7. Is mixing always energetically favourable in the stable region?
- 8. In the unstable region, a system is unstable to density fluctuations. What determines the size (length scale) of these fluctuations?
- 9. Explain differences between homogeneous and heterogeneous nucleation.

Glasses

- 10. Define a glass and show in a diagram the characteristics of the glass transition.
- 11. How does the configurational relaxation rate vary in the super cooled regime? Use the terms fragile and strong.
- 12. Can you give a physical significance to the parameter T₀ in the VFT-equation?

Colloids

- 13. What is a colloidal dispersion?
- 14. What is the stokes-Einstein equation?
- 15. What does a diffusion coefficient tell you?
- 16. What forces are there on an individual colloidal particle?
- 17. What forces are there between particles?
- 18. Discuss way to stabilise a dispersion. How can the interactions be tuned?

Polymers

- 19. Discuss different architectures for polymers (define atactic, isotactic and syndiotactic arrangements).
- 20. How can you characterise a polymer without details of the monomer chemistry?
- 21. What properties makes a polymer a good glass former?
- 22. What is the end-to-end distance for a polymer with no restrictions (freely jointed chains)? What can you tell about its entropy and internal energy?
- 23. What is the end to end distance for a real polymer (define a Kuhn segment)?
- 24. What is the excluded volume effect and how is it explained according to Flory?
- 25. Explain the coil-globule transition.
- 26. What is the typical temperature dependence of the Flory interaction parameter χ ?
- 27. Discuss the complex modulus G*.
- 28. Explain the time-temperature superposition.
- 29. What are the two main features of the stress relaxation function G(t)? Are there any similarities or differences between same kind of polymers with different degree of polymerisation?
- 30. Explain the tube model and the theory of reptation.
- 31. What causes the discrepancy between experiment ($\tau_T \propto N^{3.4}$) and the tube model ($\tau_T \propto N^3$) for the terminal time?
- 32. Why is there an almost purely elastic behaviour of G(t) for intermediate times?

Gels

- 33. Explain gel, chemical gel, physical gel, thermosetting gel, sol-gel glasses, vulcanised rubber, weak and strong physical gels.
- 34. What is the gel fraction?
- 35. Explain the three concentration regimes in a polymer solution (good solvent).
- 36. What can one expect about shrinking of coils above c* in good solvents?

Self assembly

- 37. What is a chain folded lamella?
- 38. Explain amphiphile and surfactant.
- 39. What determines optimum head group area of a micelle?
- 40. What kind of different shapes of aggregates are there and what determines the shape? Can you derive some of the criterions?
- 41. Why don't polymers generally mix?
- 42. What happens to a block copolymer with χ >2/N?